

## 159 PU Ph D Green Energy Technology

### 1 of 100

208 PU\_2016\_159\_E

The number of terminal carbonyl groups present in  $\text{Fe}_2(\text{CO})_9$  is:-

- 3
- 5
- 6
- 2

### 2 of 100

134 PU\_2016\_159\_E

For all real numbers  $x, y$  the expression  $\frac{x+y+|x-y|}{2}$  is equal to (\*):-

- the maximum of  $x$  and  $y$
- the minimum of  $x$  and  $y$
- The average of  $|x|$  and  $|y|$
- $|x + y|$

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143 PU\_2016\_159\_E

Which phenomena causes the polarization of light:-

- Reflection
- Double reflection
- Double refraction
- Diffraction

### 4 of 100

171 PU\_2016\_159\_E

In XRD analysis, determination of crystalline size is limited by the:-

- Crystalline absorption coefficient
- Lattice points
- Lattice absorption coefficient
- Mass absorption Coefficient

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146 PU\_2016\_159\_E

The viscosity of gas is directly proportional to:-

- temperature
- Characteristic gas constant
- square root of temperature
- density of gas

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114 PU\_2016\_159\_E

For a system of  $m$  linear equations in  $n$  unknowns, the Cramer's rule is applicable when:-

- $m = n$  and the coefficient matrix is non-singular
- $m \neq n$  and the coefficient matrix is non-singular
- $m \neq n$
- $m = n$

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112 PU\_2016\_159\_E

If  $A = \begin{bmatrix} 5 & 0 & 2 \\ 0 & 1 & 0 \\ -4 & 0 & -1 \end{bmatrix}$  and  $I$  be  $3 \times 3$  unit matrix, If  $M = I - A$ , then rank of  $I - A$  is:-

- 0
- 3
- 1
- 2

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203 PU\_2016\_159\_E

The emf of a Daniel cell having  $0.01 \text{ M CuSO}_4$  and  $0.2 \text{ M ZnSO}_4$  solution at  $25^\circ\text{C}$  is:-

( $\text{Zn}^{2+}/\text{Zn} = -0.76 \text{ V}$  and  $\text{Cu}^{2+}/\text{Cu} = 0.34 \text{ V}$ )

- 1.10 V
- 1.032 V
- 1.129 V
- 1.032 V

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148 PU\_2016\_159\_E

The diffusion coefficient ( $D$ ) of atoms with respect to temperature can be defined as:-

- $D = K_0 \exp(-\Delta W/KT)$
- $D = T_0 \exp(-\Delta W/KT)$
- $D = W_0 \exp(-\Delta W/KT)$
- $D = D_0 \exp(-\Delta W/KT)$

**10 of 100**

172 PU\_2016\_159\_E

Quantum confinement in solid occurs, when the size shrinks below:-

- Einstein wavelength
- Mean free path
- Green wavelength
- De-Broglie wavelength

**11 of 100**

190 PU\_2016\_159\_E

The number of unpaired electrons in  $d^6$ , low spin octahedral complex is:-

- 4
- 1
- 3
- 0

**12 of 100**

192 PU\_2016\_159\_E

The compound which obeys 18-electron rule is:-

- $\text{Cr}(\text{CO})_6$
- $\text{Mn}(\text{CO})_5$
- $\text{V}(\text{CO})_6$
- $\text{Fe}(\text{CO})_4$

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191 PU\_2016\_159\_E

In the following incomplete nuclear equation  ${}_{29}^{64}\text{Cu} \longrightarrow ? + {}_{28}^{64}\text{Ni}$ , the missing term is:-

- A positron
- A neutron
- An electron

- A proton

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131 PU\_2016\_159\_E

If  $f(1) = 2$  and  $f(x) = f(n-1) + 1/2$  for all integers  $n > 1$ , then  $F(101)$  is (\*):-

- 50  
 52  
 49  
 51

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103 PU\_2016\_159\_E

The number of ways in which 6 men and 5 women can dine at a round table if no two women are to sit together is:-

- $7! * 5!$   
 30  
  $5! * 4!$   
  $6! * 5!$

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170 PU\_2016\_159\_E

Thermoelectric generator works on the principle of:-

- Meissner effect  
 Thompson effect  
 Peltier effect  
 Seebeck Effect

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164 PU\_2016\_159\_E

The difference between the DC and AC power is arrived using:-

- $V^2R$   
  $\cos\phi$   
  $\cos\omega t$   
  $\sin\omega t$

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The Load resistance of a Solar cell is defined from the ration between:-

- Maximum output voltage and maximum output voltage  
 Maximum output current and the maximum output voltage

- Maximum output voltage and short circuit current
- Output current and the output voltage

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133 PU\_2016\_159\_E

$\int_0^1 \frac{x}{1+x^2} dx$  is:-

- $\pi/4$
- $\log\sqrt{2}$
- $\log(2)$
- 1

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176 PU\_2016\_159\_E

A monostable multivibrator circuit:-

- deliver two outputs
- Store energy
- Returns to its stand by state automatically
- Has no stable state

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194 PU\_2016\_159\_E

The 'Strainless theory' for the stability of cyclic compounds was postulated by:-

- Sachse-Mohr
- Baeyer
- Ingold
- Robinson

**22 of 100**

102 PU\_2016\_159\_E

A student is to answer 10 out of 13 questions in an examination such that he must choose at least 4 from the first five questions. The number of ways he can choose the question is:-

- 140
- 346
- 280
- 196

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141 PU\_2016\_159\_E

The relationship between the orbital quantum number "l" and the azimuthal quantum number  $n_\phi$  is:-

- l =  $n_\phi + 1$
- l =  $n_\phi$
- l =  $n_\phi - 1$
- l =  $n_\phi + (1/2)$

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116 PU\_2016\_159\_E

Which one is not true for the curve  $y = a(x-n)^2$

- horizontal line  $y = n$  is an axis of symmetry
- Represent a parabola
- for  $a > 0$  has a minimum  $y = 0$  at  $x = n$
- vertical line  $x = n$  is an axis of symmetry

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117 PU\_2016\_159\_E

The equation of a straight line that passes through point A(1,-1) and has a slope equation to -1 is:-

- $y = x$
- $y = -x$
- $y = x+1$
- $y = 1/x$

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175 PU\_2016\_159\_E

Isotopes of an atom differ with:-

- Difference in their Protons
- Difference in their Neutrons
- Difference in their electrons
- None of the above

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104 PU\_2016\_159\_E

The area of the region bounded by the curves  $y = |x-2|$ ,  $x = 1$ ,  $x = 3$  and the x-axis is:-

- 3
- 2
- 1
- 4

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147 PU\_2016\_159\_E

When the acceleration due to gravity at the surface of Earth is "g", the potential energy gain of a mass "m" raised to the radius of the Earth "R" can be:-

- $\frac{1}{4} mgR$
- $mgR$
- $2mgR$
- $\frac{1}{2} mgR$

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132 PU\_2016\_159\_E

If  $\begin{bmatrix} a & -b \\ b & a \end{bmatrix}$  is invertible under matrix multiplication then its inverse is:-

- $\frac{1}{a^2+b^2} \begin{bmatrix} a & b \\ -b & a \end{bmatrix}$
- $\begin{bmatrix} a & -b \\ b & a \end{bmatrix}$
- $\frac{1}{a^2+b^2} \begin{bmatrix} a & -b \\ b & a \end{bmatrix}$
- $\begin{bmatrix} a & b \\ -b & a \end{bmatrix}$

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142 PU\_2016\_159\_E

The resolving power of microscope is :-

- Unlimited
- Limited by the wavelength of light used
- Limited by the diameter of objective lens
- Limited by the kind of glass used

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201 PU\_2016\_159\_E

The splitting of energy levels in the presence of an external electric field is\_\_\_\_\_.

- Zeeman effect
- Kerr effect
- Compton effect
- Stark effect

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145 PU\_2016\_159\_E

The angular velocity of the rotating body is,

- $\omega = \theta/t$
- $\omega = \theta/t\sin\theta$
- $\omega = 2\pi r/t$
- $\omega = 2\pi r/ t\sin\theta$

**33 of 100**

113 PU\_2016\_159\_E

A unit matrix of order n is of rank:-

- 0
- 1
- n
- 2n

**34 of 100**

144 PU\_2016\_159\_E

The type of wave that carries sound in air is :-

- Transverse wave
- Longitudinal wave
- Electromagnetic wave
- Transverse and longitudinal waves

**35 of 100**

101 PU\_2016\_159\_E

The remainder when  $2x^3+x^2-1$  is divided by  $(x-2)$  is:-

- 5
- 19
- 13
- 9

**36 of 100**

163 PU\_2016\_159\_E

In Rayleigh scattering is caused by the.:-

- Refraction
- Reflection
- Difference in the Air Mass density
- Flow of wind

**37 of 100**

200 PU\_2016\_159\_E



Acetylene has the point group:-

- $D_{\infty h}$
- $C_{\infty v}$
- $C_{2h}$
- $C_{2v}$

**38 of 100**

100 PU\_2016\_159\_E

The quadratic equation  $4kx^2 - 8x + k = 0$  has equal roots. Then the value of k is:-

- 2
- 0.5
- 4
- 1

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149 PU\_2016\_159\_E

In MOSFET, the oxide field strength is defined as:-

- $E(x) = V_G - V_c(x)/t$
- $E(x) = V_s - V_d(x)/t$
- $E(x) = V_d - V_c(x)/t$
- $E(x) = V_d - V_s(x)/t$

**40 of 100**

205 PU\_2016\_159\_E

Among the following, the complex used for cancer chemotherapy is\_\_\_\_\_.

- $[\text{Pt}(\text{NH}_3)_4]^{2+}$
- $[\text{Pt}(\text{Cl})_4]^{2-}$
- cis- $[\text{PtCl}_2(\text{NH}_3)_2]$
- trans- $[\text{PtCl}_2(\text{NH}_3)_2]$

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118 PU\_2016\_159\_E

Root of the equation  $x^2 + ix + 2 = 0$ , where  $i = \sqrt{-1}$  is:-

- no root exist
- (i, 1)
- (-2i, i)
- (-1, 1)

**42 of 100**

111 PU\_2016\_159\_E

A skew symmetric matrix cannot be of rank:-

- 1
- 0
- greater than 1
- 1

**43 of 100**

110 PU\_2016\_159\_E

If there exist a non-zero minor of order r, then rank of A is:-

- greater than or equal to r
- less than r
- less than or equal to r
- Equal to r

**44 of 100**

161 PU\_2016\_159\_E

The spontaneous emission in LASER is achieved by:-

- Quantum well layers
- Super lattice
- Optical cavity
- Ohmic contacts

**45 of 100**

162 PU\_2016\_159\_E

3Dimensional confinement is achieved using:-

- Quantum rod
- Optical cavity
- Quantum Dot
- Quantum Well

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115 PU\_2016\_159\_E

Derivative of  $y=2^x$  is:-

- $\frac{dy}{dx} = x 2^{x-1}$
- $\frac{dy}{dx} = 2.3 \log 2 \cdot 2^x$
- $\frac{dy}{dx} = -x 2^{x-1}$

$\frac{dy}{dx} = \frac{2^{x-1}}{x}$

**47 of 100**

174 PU\_2016\_159\_E

The difference in carrier density causes:-

- Carrier tunneling
- Carrier drift
- Carrier diffusion
- Carrier recombination

**48 of 100**

179 PU\_2016\_159\_E

The plants convert the poly sac rides in to cellulose by activating:-

- Covalent bond
- Metallic Bond
- Hydrogen bond
- None of the above

**49 of 100**

207 PU\_2016\_159\_E

The IUPAC nomenclature of  $K[PCl_6]$  is:-

- potassium hexachlorophosphite(V)
- potassium hexachlorophosphate(V)
- potassium hexachlorophosphine(V)
- potassium hexachlorophosphine

**50 of 100**

204 PU\_2016\_159\_E

Activation energy can be determined from a plot of\_\_\_\_\_.

- k Vs. (1/T)
- Log k Vs. (1/T)
- k vs. T
- log k Vs. T

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202 PU\_2016\_159\_E

The Henderson equation is\_\_\_\_\_.

- $pH = pK_a + \log[\text{Salt}]/[\text{Acid}]$
- $pH = pK_a - \log[\text{Salt}]/[\text{Acid}]$

- pH =  $-\log[\text{H}_3\text{O}^+]$
- pH =  $\text{pK}_a + \log[\text{Acid}]/[\text{Salt}]$

**52 of 100**

209 PU\_2016\_159\_E

Alkali metals in liquid ammonia are blue in colour because:-

- Solvated electrons
- An ion pair is formed
- They contain alkali metal cations
- An amide ion formed

**53 of 100**

119 PU\_2016\_159\_E

For joint probability  $P(A \cap B)$  for two events A and B:-

- $P(A \cap B) = P(A)P(B) - P(A \cup B)$
- $P(A \cap B) = P(A) + P(B) - P(A \cup B)$
- $P(A \cap B) = P(A) + P(B) + P(A \cup B)$
- $P(A \cap B) = P(A) + P(B)$

**54 of 100**

130 PU\_2016\_159\_E

If  $f(x) = \int_e^x \log t \, dt$  for all position x, the  $f'(x)$  is:-

- $x \log(x)$
- $\log(x)$
- x
- $1/x$

**55 of 100**

193 PU\_2016\_159\_E

Which of the following does not possess any element of symmetry?

- (+) tartaric acid
- Ethane
- Carbon tetrachloride
- Mesotartaric acid

**56 of 100**

160 PU\_2016\_159\_E

The refractive index of material is the ratio between:-

- Speed of light in vacuum/ speed of light in air
- Speed of light in vacuum/ speed of light in material
- Speed of sound/ Speed of light
- Speed of light in water/ speed of light in air

**57 of 100**

140 PU\_2016\_159\_E

Raman Effect is due to the collision of \_\_\_\_\_.

- Photon with Electron
- Electron with atom
- Photon with molecule
- Electron with photon

**58 of 100**

177 PU\_2016\_159\_E

The sweep speed of a simple RC circuit is given by:-

- $I.t/C$
- $T_s/RC$
- $V.T_s/RC$
- $I/C$

**59 of 100**

206 PU\_2016\_159\_E

The elements  ${}_{14}^{30}\text{Si}$ ,  ${}_{15}^{31}\text{P}$ , and  ${}_{16}^{32}\text{S}$  are called\_\_\_\_\_.

- Isotopes
- Isotones
- Isomers
- Isobars

**60 of 100**

173 PU\_2016\_159\_E

The mean free path ( $\lambda$ ) increases with:-

- Increase in pressure
- Increase in collision
- Increase in temperature
- Increase in vacuum

**61 of 100**

245 PU\_2016\_159\_M

Synthesis of glucose from non-carbohydrate precursors is called:-

- Glycolysis
- Gluconeogenesis
- Glycosylation
- Saccharification

**62 of 100**

226 PU\_2016\_159\_M

Link between Glycolysis and Krebs Cycle is:-

- Oxaloacetate
- Citric Acid
- Pyruvic Acid
- Acetyl CoA

**63 of 100**

229 PU\_2016\_159\_M

The species in which the evolutionary process has been influenced by man to meet his needs is called:-

- Adventive species
- Domesticated species
- Introduced species
- Neutralized species

**64 of 100**

246 PU\_2016\_159\_M

Molecule that promote release of O<sub>2</sub> from hemoglobin is:-

- 2,3-mercaptoethanol
- Acetylcholine
- 3,2 - hemoglycerate
- 2,3 - bisphosphoglycerate

**65 of 100**

225 PU\_2016\_159\_M

Formation of glucose from source other than CO<sub>2</sub> is known as:-

- Glycolysis
- Gluconeogenesis
- Hydrogenesis
- Hydrolysis

**66 of 100**

239 PU\_2016\_159\_M

Mitotic cycle is initiated by the activation of:-

- Tubulin protein
- RNA polymerase
- MPF protein kinase
- Kinotochore protein

**67 of 100**

238 PU\_2016\_159\_M

Acid rain occurs because of pollution of air by:-

- Chlorine
- Carbon Monoxide
- CO<sub>2</sub>
- SO<sub>2</sub>

**68 of 100**

236 PU\_2016\_159\_M

Heteroblastic development is a characteristic feature of:-

- Submerged aquatic plants
- Free-floating aquatic plants
- Emergent aquatic plant
- All aquatic plants

**69 of 100**

247 PU\_2016\_159\_M

Which of the following have more energy yield per unit mass?

- Proteins
- Fatty acids
- DNA
- Carbohydrate

**70 of 100**

228 PU\_2016\_159\_M

Selectable marker that provides resistance to the antibiotic Kanamycin is:-

- Neomycin phosphotransferase
- Streptomycin phosphotransferase
- Hygromycinphosphotransferase
- Gentamycin acetyletransferase

**71 of 100**

227 PU\_2016\_159\_M

Increased melting temperature for a double strand DNA results from high content of:-

- Cytosine+Guanine
- Adenine+ guanine
- Adenine+Cytosine
- Cytocine+Thymine

**72 of 100**

259 PU\_2016\_159\_M

In protein secondary structure, the electrostatic interaction between two ionic group of opposite charges are referred as:-

- Hydrogen bonds
- Van der Waals bond
- disulfide bonds
- Salt bridges

**73 of 100**

249 PU\_2016\_159\_M

Single strand DNA can self-bind to create type of secondary structures called:-

- $\alpha$  helix &  $\beta$  sheets
- Bubbles and knots
- Hairpin & loops
- minor grooves and double helix

**74 of 100**

255 PU\_2016\_159\_M

Identify the second messenger molecule from the following:-

- cyclic adenosine monophosphate;
- adenylcyclase;
- G protein;
- phospholipase

**75 of 100**

256 PU\_2016\_159\_M

Adipose tissue stores:-

- Proteins
- Starch
- Triacyleglycerol
- Carbohydrates

**76 of 100**



237 PU\_2016\_159\_M

The eye spot or stigma perform the function of:-

- Photosynthesis
- Visibility
- Photosensitization
- Respiration

**77 of 100**

248 PU\_2016\_159\_M

The Antigen Presenting Cell among the following is:-

- Macrophage
- Monocytes
- T cell
- thymus epithelial cells

**78 of 100**

257 PU\_2016\_159\_M

Typical denaturization temperature in a PCR is:-

- 95 °C
- 37 °C
- 25 °C
- 65 °C

**79 of 100**

258 PU\_2016\_159\_M

Approximate end-to-end distance of  $3 \times 10^4$  base pair DNA is:-

- 3  $\mu\text{m}$
- 100  $\mu\text{m}$
- 30  $\mu\text{m}$
- 10  $\mu\text{m}$

**80 of 100**

235 PU\_2016\_159\_M

Agar Agar is extracted mostly from:-

- Agaricus spp
- Phaeophyceae spp
- Argemones spp
- Rhodophyceae spp

**81 of 100**

268 PU\_2016\_159\_D

In an axial flow impulse turbine, energy transfer takes place due to:-

- change in pressure energy
- change in energy because of centrifugal force
- change in relative kinetic energy
- change in absolute kinetic energy

**82 of 100**

297 PU\_2016\_159\_D

During which of the following process heat rejection takes place in Carnot cycle?

- Isothermal expansion
- Isothermal compression
- Isentropic expansion
- Isentropic compression

**83 of 100**

277 PU\_2016\_159\_D

Dissipation factor,  $\tan \delta$ , of a capacity measure by which bridge?

- Schering bridge
- Anderson bridge
- Hay Bridge
- Wien bridge

**84 of 100**

267 PU\_2016\_159\_D

An isolated system is one in which:-

- both energy and mass cross the boundaries of the system
- neither mass nor energy crosses the boundaries of the system
- mass does not cross boundaries of the system, though energy may do so
- mass crosses the boundary but not the energy

**85 of 100**

295 PU\_2016\_159\_D

A closed system is one in which:-

- mass does not cross boundaries of the system, though energy may do so
- mass crosses the boundary but not the energy
- neither mass nor energy cross the boundaries of the system
- both energy and mass cross the boundaries of the system

**86 of 100**

275 PU\_2016\_159\_D

The overall heat transfer coefficient is the:-

- resistance due to wall material
- sum of conductances
- sum of resistances
- sum of convection coefficients

**87 of 100**

285 PU\_2016\_159\_D

At thermal equilibrium:-

- absorptivity is lesser than emissivity
- sum of absorptivity and emissivity is unity
- absorptivity is equal to emissivity
- absorptivity is greater than emissivity

**88 of 100**

299 PU\_2016\_159\_D

The measurement of a thermodynamic property known as temperature is based on:-

- none of these
- Zeroth law of thermodynamics
- Second law of thermodynamics
- First law of thermodynamics

**89 of 100**

278 PU\_2016\_159\_D

If the enthalpy drop in the moving blades and fixed blades of a steam turbine is 10 kJ/kg and 15 kJ/kg respectively then what is the degree of reaction?

- 40%
- 33%
- 60%
- 67%

**90 of 100**

298 PU\_2016\_159\_D

Otto cycle is also known as:-

- constant temperature cycle
- constant volume cycle
- constant temperature and pressure cycle
- constant pressure cycle

**91 of 100**

287 PU\_2016\_159\_D

Intensive property of a system is one whose value:-

- remains constant
- depends on the mass of the system like volume
- does not depend on the mass of the system, like temperature, pressure, etc.
- is not dependent on the path followed but on the state

**92 of 100**

289 PU\_2016\_159\_D

Which one of the following is NOT an accessory for the boiler?

- Condenser
- Economizer
- Air preheater
- Feed water pump

**93 of 100**

279 PU\_2016\_159\_D

A penstock pipe of 10 cm diameter carries water under a pressure head of 100 m. If the wall thickness is 9 mm, what is the tensile stress in the pipe wall in MPa?

- 272.5
- 2725
- 545
- 1090

**94 of 100**

265 PU\_2016\_159\_D

The ratio of maximum demand of the plant to the sum of individual maximum demand of various equipments is called:-

- demand factor
- diversity factor
- load factor
- maximum demand

**95 of 100**

266 PU\_2016\_159\_D

Specific ratio for a blower is:-

- 1.20 to 1.32
- 1.11 to 1.20
- 0 to 1.11

- more than 1.32

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276 PU\_2016\_159\_D

Fin effectiveness will be increased more by:-

- higher thermal conductivity
- longer circumference
- having higher value of convection coefficient
- higher sectional area

### 97 of 100

286 PU\_2016\_159\_D

Superheated vapour behaves as:-

- exactly as gas
- approximately as a gas
- as steam
- as ordinary vapour

### 98 of 100

269 PU\_2016\_159\_D

If the distribution voltage is raised from 11 KV to 33 KV, the line power loss would be lowered by a factor:-

- 1/9
- 1/3
- 3
- 9

### 99 of 100

288 PU\_2016\_159\_D

A power station's plant load factor is defined as the ratio of:-

- average load to peak load
- minimum load to peak load
- minimum load to average load
- the energy generated to that of maximum energy

### 100 of 100

296 PU\_2016\_159\_D

For a perfect gas, according to Boyle's law (where  $p$  = Absolute pressure,  $v$  = Volume, and  $T$  = Absolute temperature):-

- $v/T = \text{constant}$ , if  $p$  is kept constant
- $p v = \text{constant}$ , if  $T$  is kept constant

- $p/T = \text{constant}$ , if  $v$  is kept constant
- $T/p = \text{constant}$ , if  $v$  is kept constant